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## Stemoninine-type alkaloids from the roots of *Stemona sessilifolia*

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Two new stemoninine-type alkaloids, stemoninine A (**1**) and B (**2**), along with two known *Stemona* alkaloids, bisdehydrostemoninine (**3**) and bisdehydrostemoninine A (**4**), were isolated from the roots of *Stemona sessilifolia*, and their structures and relative configurations were determined on the basis of spectrometric data analysis.

**Keywords:** Alkaloids; Stemoninine A; Stemoninine B; *Stemona sessilifolia*

### 1. Introduction

The roots of *Stemona sessilifolia* Miq. (Stemonaceae), named “Bai Bu” in traditional Chinese herb medicine, have been used as an antitussive agent and insecticide in China for a long time. Phytochemical studies of Stemonaceae plants have been focused thus far on *Stemona* alkaloids, and more than 90 alkaloids have been reported in the literature.<sup>1</sup> We have previously reported four sessilistemonamine-type alkaloids, with a novel tetracyclic skeleton derived from pyrrolo[1,2- $\alpha$ ]azepine, isolated from the roots of *S. sessilifolia*.<sup>2,3</sup> In our further studies on the constituents of *S. sessilifolia*, two new stemoninine-type alkaloids, named stemoninines A (**1**) and B (**2**), along with two known *Stemona* alkaloids, bisdehydrostemoninine (**3**) and bisdehydrostemoninine A (**4**),<sup>4</sup> were isolated from an ethanolic extract of the roots of *S. sessilifolia*. Here we report the isolation and characterization of the four compounds (Figure 1).

### 2. Results and discussion

Compound **1** was obtained as yellow gum. Its molecular formula was determined as C<sub>22</sub>H<sub>29</sub>O<sub>5</sub>N ( $m/z$  387.2037 [M]<sup>+</sup>) on the basis of HREI-MS. EI-MS showed the molecular ion at  $m/z$  387 and a strong fragment ion peak at  $m/z$  343 [M - CO<sub>2</sub>]<sup>+</sup>. The <sup>13</sup>C NMR and DEPT spectra of **1** showed 22 carbon atoms, including five quarternary, eight methine, six methylene and three methyl carbons. The low-field carbonyl carbons at  $\delta$  178.9, combined with the strong IR absorption bond at 1763 cm<sup>-1</sup>, suggested the presence of lactone moieties. The <sup>1</sup>H NMR spectrum of **1** showed three methyl groups at  $\delta$  1.35 (d,  $J$  = 7.0 Hz, Me-22), 1.30 (d,  $J$  = 7.0 Hz, Me-15) and 1.06 (t,  $J$  = 7.5 Hz, Me-17); two olefinic protons at  $\delta$  6.12 (d,  $J$  = 3.5 Hz, H-2) and 5.88

(d,  $J$  = 3.5 Hz, H-1); two CH groups next to an *O*-atom and a CH<sub>2</sub> group next to a *N*-atom at  $\delta$  5.38 (dd,  $J$  = 5.5, 11.0 Hz, H-18), 3.55 (td,  $J$  = 10.0, 3.5 Hz, H-8), 4.26 (dd,  $J$  = 5.5, 14.0 Hz, H-5 $\alpha$ ) and 3.73 (dd,  $J$  = 11.0, 14.0 Hz, H-5 $\beta$ ).

A spin system involving H-18, H-19, H-20 and H-22 in the <sup>1</sup>H-<sup>1</sup>H COSY experiment, coupled with the correlations between H-22 and C-21; H-20 and C-21 in the HMBC spectrum, revealed the presence of a lactone moiety comprising C-18, C-19, C-20, C-21 and C-22, which was determined to be localised at C-3 by the long-range correlations between H-18 and C-2, C-3 in the HMBC experiment. Another lactone moiety was determined by the <sup>1</sup>H-<sup>1</sup>H correlations between H-15, H-13 and H-12 in the <sup>1</sup>H-<sup>1</sup>H COSY spectrum, as well as the long-range correlations between H-15 and C-12, C-13 and C-14; H-12 and C-11 in the HMBC spectrum. It was determined to be connected with C-10 by the long-range correlations between H-10 and C-11; H-12 and C-10. The C-11 was determined to be a spiro-atom connected to two oxygen atoms by the chemical shift at  $\delta_C$  113.9 in the <sup>13</sup>C NMR spectrum. Therefore, the evidence revealed that **1** contained a basic skeleton of stemoninines.<sup>5</sup> By comparison of its <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra with those of stemoninines,<sup>6</sup> **1** has a saturated ring D and a dehydroazaazulene ring (rings A and B). The <sup>1</sup>H-<sup>1</sup>H COSY and key HMBC correlations of **1** are shown in Figure 2.

The relative configuration of **1** was revealed by NOE difference spectrum coupled with the software package of CambridgeSoft Chem3D (see Figure 3). According to a conventional  $\beta$  orientation of H-18 in most of *Stemona* alkaloids<sup>1</sup>, the correlation of H-18 with H-20 indicated the  $\beta$ -orientation of H-20. Because of the chair-conformation of the seven-membered ring B, H-5 $\alpha$

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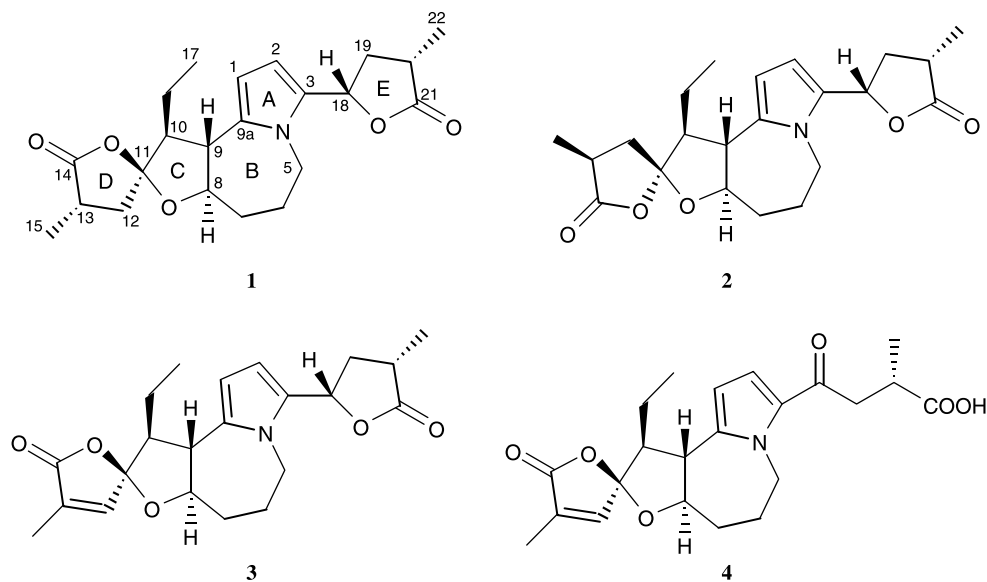


Figure 1. Structures of compounds **1**–**4**.

showed a NOE correlation with H-18. The H-9 was confirmed with a  $\beta$ -orientation by its correlation with H-5 $\beta$ . The facts of no NOE correlation between H-9 and H-8, and a visible correlation between H-8 and H-10 revealed the  $\alpha$ -orientation of both H-8 and H-10. The configuration of C-11 was determined by the fact that no NOE correlation was observed between H-9 and H-12. The  $\alpha$ -orientation of H-15 was deduced by a weak NOE correlation between H-15 and H-17.

Compound **2** was obtained as colourless prisms. Its molecular formula was determined as  $C_{22}H_{29}O_5N$ , identical to that of **1**, on the basis of HREI-MS ( $m/z$  387.2068  $[M]^+$ ). Careful analysis of spectrometric data resulted in the conclusion that **2** was a stereoisomer of **1**. The relative configuration of **2** was revealed by NOE difference spectrum coupled with the software package of CambridgeSoft Chem3D. Configurations of C-8, C-9, C-10, C-18 and C-20 proved to be identical to those of **1**. The key NOE correlations are shown in Figure 3. The configuration of C-11 was determined to be contrary

to that of **1** by the facts that a correlation between H-9 and H-12 could be observed and no correlation was observed between H-10 and H-12. The  $\beta$ -orientation of H-15 was revealed by the fact that no NOE correlation was observed between H-15 and H-17.

Compounds **3** and **4** were identified as bisdehydrostemoninine and bisdehydrostemoninine A, respectively, by comparison of their NMR spectroscopic data with those in the literature, which were isolated from *Stemona tuberosa*<sup>4</sup>. However, the mp 180–181°C and  $[\alpha]_D^{20} - 125^\circ$  ( $c = 0.11$ , MeOH) values of **3** are quite different from those reported in the literature.

### 3. Experimental

#### 3.1 General experimental procedures

Melting points were measured on a XT<sub>4</sub>-100  $\times$  microscopic melting point apparatus and are uncorrected. Optical rotations were determined on a Perkin-Elmer 343 digital polarimeter. IR spectra were obtained on a Nicolet IMPACT-400 FT-IR spectrometer. EI-MS and HREI-MS spectra were recorded on an Autospec-UltimaETOF mass spectrometer. NMR spectra were recorded on an INOVA 500 NMR spectrometer. In <sup>1</sup>H NMR spectra, the chemical shifts are given in  $\delta$  (ppm) relative to the resonances of CHCl<sub>3</sub> at  $\delta$  7.26. In <sup>13</sup>C NMR spectra, the chemical shifts are given in  $\delta$  (ppm) relative to the resonances of CHCl<sub>3</sub> at  $\delta$  77.0.

#### 3.2 Plant material

The roots of *Stemona sessilifolia* were obtained from Ding Xian market, Hebei Province, China, in June 2005.

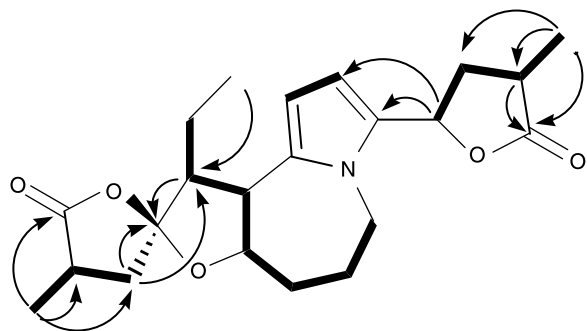
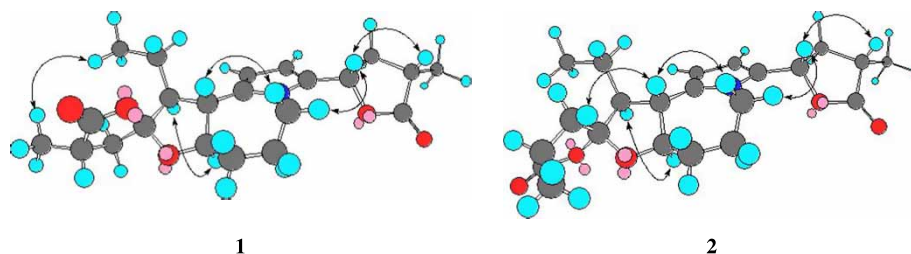


Figure 2. <sup>1</sup>H–<sup>1</sup>H correlations (bold lines) and key HMBC correlations of **1**.

Figure 3. Key NOE correlations of compounds **1** and **2**.

The plant material was identified by Professor Lin Ma of the Institute of Materia Medica. A voucher specimen (No. pc337-07) has been deposited in the Department of Natural Medicinal Chemistry, Institute of Materia Medica, Chinese Academy of Medical Sciences and Peking Union Medical College, Beijing, China.

### 3.3 Extraction and isolation

The dried roots (24.4 kg) of *S. sessilifolia* were ground and extracted with 95% EtOH at reflux. After evaporation of the solvent, a dark residue was obtained, which was suspended in 85% aq. EtOH and extracted with petroleum ether (PE; bp 60–90°C) to remove lipophilic substances. The ethanolic phase was evaporated in vacuum, and the residue was partitioned between

EtOAc and H<sub>2</sub>O. The EtOAc phase was combined and evaporated *in vacuo* to give a residue (115 g), which was subjected to silica gel column chromatography using gradient elution from 100% CHCl<sub>3</sub> to 10% MeOH/CHCl<sub>3</sub>. The fraction (41 g) eluted with CHCl<sub>3</sub> was further chromatographed on silica gel, eluting with a gradient petroleum ether/EtOAc to afford compounds **2** (9.5 mg), **3** (35.4 mg) and **4** (16.0 mg). The fraction (21 g) eluted with CHCl<sub>3</sub>/MeOH (50:1) was further chromatographed on silica gel, eluting with a gradient petroleum ether/EtOAc to afford compound **1** (8.6 mg).

#### 3.3.1 Compound 1

Yellow gum,  $[\alpha]_D^{20} -88.3$  (*c* 0.06, MeOH), IR  $\nu_{\max}$  (Microscope transmission) 2936, 1763, 1166, 965 cm<sup>-1</sup>; <sup>1</sup>H NMR and <sup>13</sup>C NMR (CDCl<sub>3</sub>) spectra data: see Table 1;

Table 1. <sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectral data for compounds **1** and **2** ( $\delta$  in ppm, *J* in Hz, in CDCl<sub>3</sub>).

Position	Stemoninine A ( <b>1</b> )		Stemoninine B ( <b>2</b> )	
	<sup>1</sup> H	<sup>13</sup> C	<sup>1</sup> H	<sup>13</sup> C
1	5.88 (d, <i>J</i> = 3.5)	103.0 (d)	5.94 (d, <i>J</i> = 3.5)	104.3 (d)
2	6.12 (d, <i>J</i> = 3.5)	106.8 (d)	6.12 (d, <i>J</i> = 3.5)	106.9 (d)
3		128.5 (s)		128.5 (s)
5	4.26 (dd, <i>J</i> = 5.5, 14.0, H <sub>α</sub> ) 3.73 (dd, <i>J</i> = 11.0, 14.0, H <sub>β</sub> )	45.2 (t)	4.29 (dd, <i>J</i> = 5.5, 14.0, H <sub>α</sub> ) 3.70 (m, H <sub>β</sub> )	45.4 (t)
6	2.05, 1.63 (each m, 1 H)	26.2 (t)	2.05, 1.62 (each m, 1 H)	25.9 (t)
7	2.27 (td, <i>J</i> = 11.5, 3.5, H <sub>α</sub> ) 1.72 (m, H <sub>β</sub> )	35.5 (t)	2.33 (m, H <sub>α</sub> ) 1.60 (m, H <sub>β</sub> )	34.4 (t)
8	3.55 (td, <i>J</i> = 10.0, 3.5)	83.4 (d)	3.67 (m)	80.9 (d)
9	3.13 (t, <i>J</i> = 11.0)	47.6 (d)	2.74 (t, <i>J</i> = 10.0)	49.8 (d)
9a		133.8 (s)		133.7 (s)
10	2.42 (m)	50.2 (d)	2.86 (dd, <i>J</i> = 7.5, 9.5)	50.1 (d)
11		113.9 (s)		115.0 (s)
12	2.48 (dd, <i>J</i> = 8.5, 13.5) 2.12 (t, <i>J</i> = 13.5)	40.7 (d)	2.35, 2.07 (each m, 1 H)	38.5 (d)
13	2.99 (m)	34.4 (d)	2.99 (m)	34.7 (d)
14		178.9 (s)		178.9 (s)
15	1.30 (d, <i>J</i> = 7.0)	15.2 (q)	1.29 (d, <i>J</i> = 7.0)	15.0 (q)
16	1.77 (m)	20.7 (t)	1.74, 1.58 (each m, 1 H)	23.6 (t)
17	1.06 (t, <i>J</i> = 7.5)	12.9 (q)	0.98 (t, <i>J</i> = 7.5)	12.3 (q)
18	5.38 (dd, <i>J</i> = 5.5, 11.0)	71.5 (d)	5.37 (dd, <i>J</i> = 5.5, 11.0)	71.7 (d)
19	2.70, 2.22 (each m, 1 H)	34.7 (t)	2.70, 2.22 (each m, 1 H)	34.7 (t)
20	2.80 (m)	36.0 (d)	2.80 (m)	36.0 (d)
21		178.9 (s)		178.9 (s)
22	1.35 (d, <i>J</i> = 7.0)	15.0 (q)	1.35 (d, <i>J</i> = 7.0)	14.9 (q)

HREI-MS  $m/z$  387.2037  $[M]^+$  (calcd for  $C_{22}H_{29}O_5N$ , 387.2046), EI-MS  $m/z$  388 (7,  $[M + H]^+$ ), 387 (88,  $M^+$ ), 358 (11), 343 (90), 340 (35), 218 (100), 174 (42), 172 (35).

### 3.3.2 Compound 2

Colourless prisms, mp 199–200°C,  $[\alpha]_D^{20}$  0 ( $c$  0.06, MeOH), IR  $\nu_{\max}$  (Microscope transmission) 2932, 1753, 1161, 956  $cm^{-1}$ ;  $^1H$  NMR and  $^{13}C$  NMR ( $CDCl_3$ ) spectra data: see Table 1; HREI-MS  $m/z$  387.2068  $[M]^+$  (calcd for  $C_{22}H_{29}O_5N$ , 387.2046), EI-MS  $m/z$  388

(24,  $[M + H]^+$ ), 387 (100,  $M^+$ ), 358 (25), 343 (85), 218 (98), 174 (37), 172 (20).

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